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# Performance of absorption coefficient measurements for the *in situ* determination of chlorophyll-a and total suspended matter

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## Abstract

The concentration of chlorophyll-a ([chl-a]) and total suspended matter ([TSM]) are important parameters in biological oceanography. [Chl-a] is a commonly used proxy for estimating phytoplankton biomass while [TSM] also includes detrital material and mineral particles and thus influences light attenuation and photosynthetic activity in the water column.

For characterizing the distribution (patchiness) of both parameters adequately over a longer time period, fast and effective measurement methods are required that can also be applied *in situ* or continuously. Thus, alternatively to direct determination of [chl-a] and [TSM], optical proxy values are often measured. The PSICAM is an integrating cavity approach for measuring absorption coefficients of water constituents with high precision which can be used also continuously (flow-through-PSICAM). In this study, the performance of these absorption measurements for [chl-a] and [TSM] determination was evaluated and compared with the performance of traditional approaches using chl-a fluorescence and turbidity measurements.

Data were collected in the German Bight (North Sea) in 2010 and 2011. For [chl-a], fluorescence measurements are compared with pigment absorption coefficient values at a

29 wavelength of 676 nm ( $a_{\Phi 676 \text{ nm}}$ ), while the [TSM]-proxies were turbidity and particle  
30 absorption at 700 nm ( $a_{p 700 \text{ nm}}$ ). As reference data, HPLC-determined [chl-a] and  
31 gravimetrically determined [TSM] were used.

32 Our results showed linear relationships between [chl-a] and fluorescence or  $a_{\Phi 676 \text{ nm}}$ ,  
33 respectively. Coefficients of determination ( $R^2$ ) were in a range of 0.71 to 0.88, with the  
34 higher values related to the absorption measurements. Furthermore, it was demonstrated  
35 that fluorescence underestimates [chl-a] depending on ambient photosynthetically active  
36 radiation (PAR). Linear relationships were also observed between [TSM] and its optical  
37 proxies with  $R^2$  values between 0.93 and 0.98. Turbidity measurements appeared to be  
38 influenced to a certain extent by the physical properties of the suspended material, resulting  
39 in a slightly higher variability than the  $a_{p 700 \text{ nm}}$  measurements.

40 Absorption measurements turned out to be promising optical proxies for determining [TSM]  
41 and [chl-a] due to their lower variability compared with the other proxies. This improved  
42 accuracy could be already partially achieved also for continuous measurements. Moreover, a  
43 combination of the different optical methods has the potential to provide additional  
44 information besides concentration, such as the source of TSM in the water or physiological  
45 condition of the phytoplankton.

46

47 **Keywords:** chlorophyll-a, total suspended matter, fluorescence, absorption, German Bight,  
48 evaluation of flow through-PSICAM

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50

## 51 **1. Introduction**

52 The North Sea is, like most shelf and coastal seas, a highly productive ecosystem fostered  
53 by relatively shallow waters and a constant supply with nutrients by river discharge  
54 (Ducrotoy, et al., 2000). The impact of e.g. fishing, shipping, pollution and energy production  
55 is high today and is likely to further increase in future. By enabling the detection of long-term  
56 changes driven by this increasing utilization as well as by climate changes, time series like

57 the Helgoland Roads (Wiltshire, et al., 2010) or the Continuous Plankton Recorder (Reid, et  
58 al., 2003) have shown their value. Information on phytoplankton, the basis of the marine food  
59 web, is of particular importance, as changes in this part of the ecosystem also affect higher  
60 trophic levels. In this context, phytoplankton biomass is a key parameter and needed for  
61 productivity and ecosystem model calculations. A commonly used and early introduced  
62 approach for its estimation is the measurement of the concentration of the major algal  
63 pigment chlorophyll-*a* (referred to as [chl-*a*] hereafter) in the water (Harvey, 1933; Mineeva,  
64 2011). Although the relationship between [chl-*a*] in the water and phytoplankton biomass is  
65 not constant but varies with species composition (Llewellyn and Gibb, 2000), acclimatization  
66 to light conditions (Falkowski and Owens, 1980; Llewellyn, et al., 2005), or developmental  
67 stage (Llewellyn and Gibb, 2000), it allows reasonable estimates about phytoplankton  
68 without tedious microscopic observations. In addition to [chl-*a*], total suspended matter  
69 concentration in the water (hereafter referred to as [TSM]) is another important parameter to  
70 measure, because it is governing e.g. the underwater light regime and thereby influencing  
71 phytoplankton photosynthetic activity.

72 Today, the most accurate method for direct [chl-*a*]-measurement is high performance liquid  
73 chromatography (HPLC), while for [TSM] it is gravimetry. Both approaches require a high  
74 degree of manual handling and are restricted to the analysis of discrete samples from fixed  
75 stations or occasional research cruises. Hence, although accurate, they often provide only  
76 insufficiently detailed information because of the lack in spatial and temporal resolution.

77 Furthermore, the high effort in the analyses hampers the establishment and maintenance of  
78 time series covering these parameters.

79 Measurement of optical proxy values can be an alternative to the measurement of the  
80 parameters itself. Although less accurate than direct measurements, they offer a cost-  
81 effective, fast and convenient way to estimate a variety of parameters relevant to marine  
82 researchers, and many of the available methods can be applied on an *in situ* or continuously  
83 basis (Moore, et al., 2009). While [TSM] is traditionally determined via light scattering in the  
84 water by turbidimeters (Sternberg, et al., 1974; Vant, 1990), [chl-*a*] is commonly determined

85 via its fluorescence (Lorenzen, 1966; Ostrowska, et al., 2000). Because the intensity of the  
86 fluorescence signal depends amongst others on [chl-a], it allows estimates about the  
87 phytoplankton biomass in a given sample. Although the signal can be accurately calibrated  
88 against HPLC-determined [chl-a], the calibration is not stable over longer periods of time as  
89 well as in changing environmental conditions (Wiltshire, et al., 1998; Beutler, et al., 2002;  
90 Beutler, et al., 2004). The variability in the ratio between fluorescence and [chl-a] is caused  
91 e.g. by variations in the physiological state of the cells, in species composition, and in the  
92 variable fluorescence yield itself in dependence of the actual and short-time historic light  
93 conditions (Falkowski and Kiefer, 1985; Cunningham, 1996; Chekalyuk and Hafez, 2011).  
94 Despite this uncertainties and due to their convenience, approaches based on fluorescence  
95 were widely used for mapping of horizontal and vertical phytoplankton distributions (e.g.  
96 Uehlinger, 1985; Seppälä and Balode, 1998; Wiltshire, et al., 1998; Muylaert, et al.,  
97 2006). Due to the relative simplicity of turbidity and fluorescence measurements, the  
98 respective instruments are also predestined to be used in unattended, long term-  
99 measurement systems to further increase data resolution in time and space. For example,  
100 they are mounted on time series stations (Badewien, et al., 2009), or in FerryBoxes aboard  
101 on ships of opportunity (Fleming and Kaitala 2006; Lips, et al., 2007; Petersen, et al., 2008;  
102 Petersen, et al., 2011) to monitor TSM and phytoplankton dynamics in real-time.  
103 Alternatively to turbidity and fluorescence, [TSM] and [chl-a] can be determined by  
104 measuring absorption coefficients (Yentsch and Phinney, 1989; Bricaud, et al., 1995;  
105 Wollschläger, et al., 2013). With respect to chl-a, the advantage is given by measuring a  
106 physical parameter of the cells instead of a more physiological value as it is the case for  
107 fluorescence, making [chl-a] determination potentially more accurate. This is especially  
108 important when data were obtained from unattended devices (satellites, gliders, FerryBoxes)  
109 where direct [chl-a] measurements for calibration are not necessarily available. However,  
110 measurement of spectral absorption coefficients in seawater is more complicated than  
111 measurement of chl-a fluorescence or turbidity. With the exception of some coastal areas,  
112 the content of absorbing material in seawater is relatively low and additional light loss due to

113 scattering on particles requires the application of empirical correction procedures. Several  
114 instruments are available for the determination of absorption coefficients (Pegau, et al., 1995;  
115 Moore, et al., 2009), and some of them address the problem of low concentration by using  
116 larger linear cuvettes or by using liquid wavelength capillary cells (Kirkpatrick, et al., 2011).  
117 Nevertheless, the bias due to scattering remains. In this context, devices based upon the use  
118 of integrating cavities (Fry, et al., 1992; Kirk, 1997; Pope, et al., 2000; Röttgers, et al., 2005)  
119 are promising, as they are independent of the need for scattering corrections, highly sensitive  
120 due to a long optical path length, and still compact in design. Recently, efforts have been  
121 made to adapt these systems also for continuous measurements and tests *in vitro* (Musser,  
122 et al., 2009) and *in situ* (Wollschläger, et al., 2013) have been performed.

123 In this study, we evaluated the performance of absorption coefficient measurements for the  
124 determination of [TSM] and [chl-*a*] by comparison with traditional methods like fluorescence  
125 and turbidity measurements. Because of the aforementioned advantages, the absorption  
126 coefficients were determined using an integrating cavity approach, a point source integrating  
127 cavity absorption meter (PSICAM). As mentioned, the possibility of obtaining continuous  
128 measurements is crucial; for this reason, also the performance of a flow-through-version of  
129 this device was tested (ft-PSICAM). Aim was to evaluate whether potential advantages of  
130 absorption coefficient measurements for the determination of [TSM] and [ch-*a*] could also be  
131 achieved with continuous measurements.

132

133

## 134 **2. Materials and Methods**

### 135 **2.1 Discrete sampling and continuous measurements**

136 Data were collected on several cruises with the R.V. 'Heincke' in 2010 and 2011 in the  
137 German Bight, North Sea (Fig. 1; tab. 1). At the hydrographic stations (dots in figure 1),  
138 discrete samples were taken from depths of around 4 to 5 meters using a sampling rosette  
139 (SBE 32, Sea-Bird Electronic, Inc.) equipped with seven 9-L Niskin sampling bottles  
140 (Hydrobios, Germany). The water was transferred into 10-L PE containers and processed

141 subsequently for measurements of salinity, temperature, chl-*a*, turbidity, TSM, and  
142 absorption of water constituents. Continuous measurements of chl-*a* fluorescence and  
143 turbidity were performed by a FerryBox system as described in Petersen et al. (2011). The  
144 water intake for the FerryBox was in the moon pool of the research vessel at ca. 4 m depth,  
145 thus widely comparable to the sampling depth for the discrete measurements. Continuous  
146 measurements of water constituent absorption were performed on the cruises in April and  
147 June 2011 using a flow-through-PSICAM (ft-PSICAM) as described in Wollschläger et al.  
148 (2013). The water for these measurements was provided from a bypass of the FerryBox  
149 water supply. This ensured that both devices got the same water samples and the results  
150 were comparable. From the continuous datasets, values for the cruise stations were  
151 extracted according to the corresponding time stamps.

152

### 153 **2.2 Determination of chlorophyll-*a* by HPLC**

154 Concentration of chl-*a* was measured by high performance liquid chromatography (HPLC).  
155 Water samples (1-5 L) were filtered through pre-combusted GF/F filters (Whatman, USA, Ø  
156 47 mm), subsequently, the filters were shock-frozen in liquid nitrogen and stored at -80 °C.  
157 Later in the laboratory, the pigments were extracted from the filters with 100 % acetone for  
158 24 h at -30 °C (Wright, et al., 1991; modified). The extracts were transferred into 2 ml glass  
159 vials and simultaneously cleaned from particles by passing them through 0.2 µm syringe  
160 filters (regenerated cellulose, Spartan, A13). The separation and analysis of the pigments  
161 was carried out by a HPLC system from JASCO (Japan) using the method described in  
162 Zapata et al. (2000).

163

### 164 **2.3 Determination of total suspended matter by gravimetry**

165 Concentration of total suspended matter was determined by filtration of 1-8 L of the water  
166 sample through pre-combusted, pre-washed and pre-weighted GF/F filters (Whatman, USA,  
167 Ø 47 mm). Previously to usage, the filters were wet with purified water to avoid saturation of  
168 the rim with sea water during filtration. This was done in order to reduce the amount of salt

169 that cannot be washed out of the filter afterwards (Stavn, et al., 2009). Additionally at each  
170 cruise, already filtered and therefore particle-free seawater was applied on empty filters like  
171 real samples, and the average mass increase due to the remaining salt was determined. It  
172 was subtracted from the mass values of all samples of the particular cruise before calculating  
173 total suspended matter concentration gravimetrically (in mg/L; see Stavn et al. 2009 for  
174 details).

175

## 176 **2.4 Measurement of optical proxy-values**

### 177 **2.4.1 Chlorophyll-a fluorescence and turbidity**

178 Chl-a fluorescence was continuously measured with SCUFA-II-sensors (Turner Design,  
179 USA) mounted in the FerryBox system. Calibration of the sensor was performed by the  
180 manufacturer. Sensitivity was checked previously to the cruise by using a solid standard  
181 provided by the manufacturer. The same sensor was also used for continuous turbidity  
182 determination using a nephelometric method measuring light scattering at an angle of 90°.  
183 For turbidity, the sensor was calibrated before the cruises in the laboratory using different  
184 suspensions of formazine in a range of 0.5 to 35 formazine nephelometric units (FNU). The  
185 turbidity of these suspensions were determined with a calibrated Hach 2100N IS turbidimeter  
186 (Hach, USA). The linear relationship between the SCUFA-II and the laboratory instrument  
187 was strong in all cases with  $R^2 > 0.99$ . For comparison with data from discrete samples,  
188 corresponding fluorescence and turbidity values were extracted from the continuous data set  
189 based on time of the sampling. In addition, manual turbidity measurements of samples have  
190 been done onboard using a Hach 2100P ISO turbidimeter (Hach, USA).

191 In preliminary analysis, an offset was visible –despite calibration– for both the turbidity and  
192 the fluorescence data measured with the SCUFA-II. After the cruises, for correction of these  
193 offsets, purified water was measured to obtain zero-values for both parameters. These  
194 values were then subtracted from the corresponding data. Since the fluorescence  
195 measurements are temperature depended, the purified water was maintained at 5 and 15 °C,  
196 because these temperatures were representative for the water measured in spring and



197 summer/autumn, respectively. For turbidity, a mean value of 0.2 nephelometric turbidity units  
198 (NTU) was measured, for fluorescence a mean value of 0.55 and 0.54 arbitrary units (AU) for  
199 5 °C and 15 °C, respectively. However, it should be noted that this correction was not  
200 sufficient for a complete elimination of the offsets.

201

#### 202 **2.4.2 Spectral absorption coefficient determination**

203 Discrete absorption measurements were made using a PSICAM as described in Röttgers et  
204 al. (2005) and Röttgers & Doerffer (2007). It is referred to as 'conventional PSICAM' in the  
205 following. The only change in the described setup was the use of a Hamamatsu C10083CA  
206 mini-spectrometer (Hamamatsu, Japan) during the cruises in April and June 2011. The  
207 theoretical background as well as the equations for the calculation of spectral absorption  
208 coefficients or cavity wall reflectivity is given elsewhere (Kirk, 1997; Leathers, et al., 2000).

209 Measurements of the cavity wall reflectivity for calibration of the PSICAM were performed  
210 daily in triplicates according to Röttgers et al. (2005) using an always freshly prepared  
211 solution of the dye nigrosine, purified water (>18.2 MΩm) as reference and 0.1 % NaOCl  
212 solution (Riedel de Haën, Germany) for cleaning. Spectral absorption coefficients of the  
213 nigrosine solution were determined in advance using liquid waveguide capillary cells (path  
214 lengths: 0.5 and 2 m; WPI, USA), again with purified water as reference. Prior to their use,  
215 both solutions were adapted to the same temperature in a water bath. Sample  
216 measurements with the conventional PSICAM were also performed in triplicate, with a  
217 reference measurement previous to each sample. Subsequent to the sample measurement,  
218 the cavity was rinsed with purified water for cleaning and then the reference was measured  
219 again. Temperature for both was recorded and used together with the determined salinity  
220 (measured with a WTW 330i portable conductivity sensor; WTW, Germany) for correction of  
221 variations in pure water absorption due to temperature and salinity differences between  
222 sample and reference (Röttgers and Doerffer, 2007).

223 Continuous measurements of spectral absorption coefficients were performed taking  
224 advantage of a custom-made ft-PSICAM. Its setup as well as the measurement, the cleaning

225 and the calibration procedure is described in Wollschläger, et al., 2013. Calibration was  
226 performed daily; sample water was measured every 5 seconds and averaged over one  
227 minute. Since reference measurements were only performed at stations (in intervals of one  
228 to eight hours), spectral absorption coefficients on a transect line were calculated using  
229 linearly interpolated reference values based on the data measured on the stations at the  
230 beginning and the end of the transect. Correction of the coefficients for the influence of salt  
231 and temperature were done using temperature and salinity data obtained from the FerryBox.  
232 To maintain the performance of the system, the ft-PSICAM was cleaned in the same way as  
233 during calibration at every station. Nevertheless, the continuously obtained absorption  
234 coefficients were distorted by small contaminations and changes in the reference signal and  
235 were therefore corrected using discrete data of the conventional PSICAM (see also  
236 Wollschläger, et al., 2013). The particulate absorption coefficient at 700 nm ( $a_{p700\text{ nm}}$ ) served  
237 as a proxy value for [TSM], while for [chl-*a*] the estimated absorption coefficient of  
238 phytoplankton pigments at 676 nm ( $a_{\Phi 676\text{ nm}}$ ) was used.  $a_{p700\text{ nm}}$  is identical with total  
239 absorption of water constituents at this wavelength, assuming the contribution of both CDOM  
240 and pigments is negligible in this region.  $a_{\Phi 676\text{ nm}}$  was estimated by subtracting  $a_{p700\text{ nm}}$  from  
241 total absorption of water constituents at 676 nm. For details, see Wollschläger, et al., 2013.

242

## 243 **2.5. Additional data sets**

244 In the evaluation of chl-*a* determination by fluorescence and absorption, additional datasets  
245 were used. The concentration of CDOM was qualitatively estimated by fluorescence using a  
246 Cyclops-7 sensor (WETLabs, USA) mounted in the FerryBox. Furthermore, for the cruises in  
247 April and June 2011, data about photosynthetic active radiation (PAR) were obtained from  
248 the hydrographic time series station 'Wattenmeer' (Badewien, et al., 2009; Reuter, et al.,  
249 2009).

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253

### 254 3. Results & Discussion

#### 255 3.1 Station measurements

256 The first step in the evaluation of the different optical proxies for [TSM] and [chl-a]  
257 determination was the comparison of the respective data obtained at the cruise stations with  
258 the corresponding reference data ([TSM] determined gravimetrically and [chl-a] measured by  
259 HPLC). Figures 2 and 4 as well as table 2 show the respective correlations. The observed  
260 values of [TSM] and [chl-a] span roughly two orders of magnitude, with the lower values  
261 sampled in the more offshore regions and the higher values near the coast. For [TSM], most  
262 of the data (77 %) were below  $5 \text{ g/m}^3$ , while for [chl-a], 87 % were below a concentration of 5  
263  $\mu\text{g/L}$ . Turbidity was determined manually using the Hach turbidimeter and continuously by  
264 the SCUFA-II. The latter was also used for chl-a fluorescence measurements. Particle  
265 absorption coefficients at 700 nm ( $a_{p\ 700\ \text{nm}}$ ) and pigment absorption coefficients at 676 nm ( $a_{\phi\ 676\ \text{nm}}$ )  
266 were measured manually with the conventional PSICAM and continuously with the ft-  
267 PSICAM.

268

##### 269 3.1.1 Total suspended matter determined at cruise stations

270 In all cases, statistically significant linear relationships between the respective optical proxy  
271 value and [TSM] were observed, but the variability in the data was different for the four  
272 methods: The strongest correlation was obtained for the PSICAM measurements of  $a_{p\ 700\ \text{nm}}$   
273 ( $R^2= 0.98$ ; Fig. 2 E), followed by the Hach turbidity measurements ( $R^2= 0.95$  Fig. 2 A) and the  
274 continuously obtained optical proxy values ( $R^2= 0.94$  for the ft-PSICAM measurements and  
275  $R^2= 0.93$  for the SCUFA-II; Fig. 2 C, G). Thus, the manually measured optical proxies  
276 allowed a slightly more stable determination of [TSM] than the continuously obtained data,  
277 what is also supported by the standard error of the slope of the respective fits given in table  
278 2. This was expected, since the continuous measurements were performed under less  
279 controlled conditions compared to the manual ones.

280 Although the higher coefficient of determination and the lower error of the slope indicate a  
281 better performance of the absorption-based approach for the manual measurements, the  
282 same was not achieved for the continuous measurements. While the  $R^2$ -values were nearly  
283 identical, the error of the slope was higher for the ft-PSICAM measurements than for those  
284 performed with the SCUFA-II. Partly, this higher variability could be a statistical effect caused  
285 by the considerably lower sample size and range in the data of the continuous absorption  
286 measurements. But logarithmic transformation of the data (Fig. 2 B, D, F, and H) revealed  
287 increased scattering around the applied linear fits below  $[TSM] < 3 \text{ g/m}^3$ . Although  
288 observable for all instruments, it was especially high for the ft-PSICAM and probably the  
289 reason for the higher error of the slope. Thus, by now, the measurements of the ft-PSICAM  
290 should be treated with caution at such low  $[TSM]$ .

291 One reason for the variable performances of the methods for  $[TSM]$ -determination are the  
292 different measurement errors of the instruments used. The relative error of the various  
293 methods in dependence of the measured  $[TSM]$  up to  $10 \text{ g/m}^3$  is shown in figure 3. For both  
294 manual and continuous turbidity measurements, the averaged error of the slope obtained  
295 from the sensor calibrations was considered as measurement error of the instruments,  
296 including errors due to manual handling (Hach turbidimeter: 0.249; SCUFA-II: 0.368). The  
297 error associated with  $a_{p, 700 \text{ nm}}$  varied with the value of the respective measurement, since it  
298 was a combination of the error in wall reflectivity determination and precision of the  
299 instrument ( $\sim 2\%$  and  $0.001 \text{ m}^{-1}$  for the conventional PSICAM; Röttgers and Doerffer, 2007,  
300 for the ft-PSICAM  $\sim 5\%$  and  $0.007 \text{ m}^{-1}$ ; unpublished data). Uncertainties in the gravimetrically  
301 determination of  $[TSM]$  itself were composed from errors in determination of the mass  
302 retained on the filter (average 0.003 mg), in the salt-offset (average 0.3 mg) as well as from  
303 errors in the determination of the filtered volume (average 0.4 %). The latter depended on the  
304 amount of TSM in the water, thus the related error was quite variable for the different  
305 samples. However, the contribution of the gravimetrically  $[TSM]$  determination to variations in  
306 the linear relationships can be considered as minor. Generally, instrumental errors increased

307 at  $[TSM] < 3 \text{ g/m}^3$  (Fig. 3), indicating that they were primarily responsible for the observed  
308 variations in the relationships between  $[TSM]$  and optical proxies at low values.  
309 In cases where instrumental errors are small, observed variability in the relationship between  
310 the respective optical proxy and  $[TSM]$  can also be caused by a different susceptibility to  
311 natural variability in the optical properties of the investigated particles. Turbidity and particle  
312 absorption are not only functions of  $[TSM]$ , but also of particle size-distribution and origin  
313 (e.g. organic or inorganic; Babin, et al., 2003; Babin and Stramski, 2004; Astoreca, et al.,  
314 2012). A possible manifestation of this effect can be seen in the data displayed in figure 2 A.  
315 Based on the calibration of the instrument, the slope of the correlation between  $[TSM]$  and  
316 turbidity was expected to be close to 1. This was not the case, apparently caused by some  
317 onshore stations of the September cruise in 2010, which showed relatively low turbidity  
318 values although there were high  $[TSM]$  present (empty circles marked with arrows). Since the  
319 relative measurement errors were generally low at high values (compare figure 3), these data  
320 points might be 'outliers' because of differing scattering properties of the TSM present at  
321 these stations. Possibly, the storm event occurring during this time caused re-suspension of  
322 larger particles than usually present, which are less efficient light scattering per mass than  
323 smaller particles (Stramski, et al., 2007; Wozniak, et al., 2010). In contrast,  $a_{p 700 \text{ nm}}$  seemed  
324 to be much less influenced by the properties of TSM, because in figure 2 E, the respective  
325 stations of September 2010 appeared not as outliers. But also in the continuously measured  
326 turbidity data, they were not as clearly distinguishable from the other stations as in the  
327 manual measurements. It can be speculated that the large particles were only aggregates of  
328 smaller ones, which were disrupted by the impeller pump of the FerryBox. This would have  
329 made the particles measured more uniform and thus masking existing anomalies in particle  
330 size distribution.

331

### 332 **3.1.2 Chlorophyll-a determined at cruise stations**

333 The correlation of  $[chl-a]$  with the measurements of the optical proxy values resulted in  
334 statistically significant linear relationships (Fig. 4, tab. 2). Despite the correction applied on

335 the fluorescence measurements (see 2.4.1), still an offset in the data remained (0.547 +/-  
336 0.092 AU). Possibly, an insufficient temperature correction of the sensor was the reason, or  
337 an incorrect separation of the excitation light from the emission light by the filter of the  
338 instrument, which created a signal even if no chlorophyll was present in the sample.  
339 However, linear correlations with [chl-*a*] were in general stronger for the absorption than for  
340 the fluorescence measurements, and as for TSM-determination, the manual measurements  
341 (Fig. 4 E) performed better than the continuous measurements (Fig. 4 G). The slightly higher  
342 error in the slope for ft-PSICAM measurements compared to those of the SCUFA-II can be  
343 related to the smaller number of measurements performed. Differences in the offset between  
344 the fits for the manual and the continuous obtained  $a_{\phi 676 \text{ nm}}$  data were only of low statistical  
345 significance ( $p= 0.08$  derived from a t-test). Probably, the higher value for the continuous  
346 measurements was caused by the lack in high [chl-*a*] measured by the ft-PSICAM, what  
347 distorted the regression line. However, considering the offsets as negligible small, the fits  
348 were forced through zero in the following. This assumption is realistic, because chlorophyll-  
349 related absorption should be zero in the absence of chlorophyll, and led to identical slopes  
350 for both PSICAMs (0.0182 +/-0.0004 for manual  $a_{\phi 676 \text{ nm}}$ , and 0.0179 +/- 0.0008 for  
351 continuous  $a_{\phi 676 \text{ nm}}$ ,  $p= 0.7$  according to a t-test). The relative error of the slope changed to  
352 2.2 % and 4.5 %, respectively. In order to emphasize the smaller values of which the dataset  
353 consists to a majority, also the logarithm of the data was taken and fitted in a linear way. In  
354 this case, the coefficient of determination for the  $a_{\phi 676 \text{ nm}}$  measurements increased (Fig. 4 F,  
355 H), while it decreased for the fluorescence measurements (Fig. 4 B) because of high  
356 scattering in the data at smaller values. In summary, the stronger linearity in the relationship  
357 obtained for both the normal and the log-data, especially at lower [chl-*a*], indicated that in  
358 general  $a_{\phi 676 \text{ nm}}$  is a more accurate proxy for [chl-*a*] than *in situ* chl-*a* fluorescence.  
359 The weaker performance of the fluorescence measurements might be caused on the one  
360 hand by an influence of water turbidity on the fluorescence signal (TurnerDesigns, 2004), but  
361 on the other hand also by its strong link to phytoplankton physiology (Falkowski and Kiefer,  
362 1985). Therefore, it is influenced by a variety of parameters, e.g. the absorption properties of

363 photosystem II, which are determined by the pigment composition of the cells (Hoepffner and  
364 Sathyendranath, 1991). Although principally species specific, this composition can also vary  
365 due to long term adaptation of the cells to certain light conditions (Yentsch and Yentsch,  
366 1979; Soohoo, et al., 1986; Lutz, et al., 2001). Additionally, self-shading of the chloroplasts,  
367 the so-called 'pigment-packaging' (Duysens, 1956; Morel and Bricaud, 1981; Kirk, 1994), has  
368 an impact on light absorption and therefore the amount of energy which is supplied to the  
369 photosystem. Furthermore, the fluorescence yield itself, which reflects the proportion of  
370 absorbed light energy that the cell uses for photochemistry, is also variable. It is discussed to  
371 be influenced by the nutrient status of the cells (Kiefer, 1973b; Parkhill, et al., 2001), as it can  
372 have an impact on the general condition of the photosynthetic apparatus and its enzymes.  
373 Other factors are non-photochemical quenching mechanisms which allow a short-term  
374 acclimatization to ambient light conditions. They can be triggered by ambient light (Szabo, et  
375 al., 2005) or a diel rhythm (Prézelin and Ley, 1980). Influence of light acclimatization on the  
376 fluorescence signals is also indicated for the data obtained in this study: When considering  
377 only the fluorescence values measured during the night, when the cells were some kind of  
378 'dark-adapted' (Fig. 4 C, D), the variability in the data was much lower compared to that  
379 observable in figure 4 A and B (the value shown in brackets was considered as an outlier  
380 and omitted from the calculation).

381 The –compared to the fluorescence signal– smaller variability of the  $a_{\Phi 676 \text{ nm}}$  measurements  
382 can be explained by the fact that they are only influenced by variations in the chl-*a* specific  
383 absorption properties, which take place on comparably long timescales. They emerge from  
384 the pigment packaging effect and the variable pigment composition of phytoplankton which  
385 also affect fluorescence (see above). However, pigment composition changes are only of  
386 importance if absorption measurements were conducted in the blue region of the light  
387 spectrum, where also other pigments absorb in addition to chl-*a*. For measurements in the  
388 red absorption maximum of chl-*a*, as were done for this study, they can be widely neglected,  
389 because only chl-*a* and –to a lesser extent– chlorophyll-*b* and phycobilines absorb there.  
390 According to HPLC measurements, chlorophyll-*b* concentration was mostly below 20 %

391 (mean value: 6 %) of that of chl-*a* in this study, and no blooms of cryptophytes or  
392 cyanobacteria containing phycobilines were observed. Thus, in most cases, the whole  
393 absorption at 676 nm can be considered due to chl-*a*, attributing nearly all natural variability  
394 observed in the relation between [chl-*a*] and  $a_{\phi 676 \text{ nm}}$  to the pigment packaging effect.  
395 For this reason, the accuracy of [chl-*a*] determination using the absorption-based approach is  
396 –with exception of instrumental errors– in the first line limited by natural variability which is  
397 inherent in the phytoplankton due to the packaging effect. Absorption is influenced by fewer  
398 factors than fluorescence, what explains the better performance of the PSICAM-based  
399 measurements.

400

### 401 **3.2. Continuous measurements**

402 In order to evaluate the continuous measurement systems not only by means of the station  
403 data, also the transect measurements were compared. Only transects of the cruises in April  
404 and June 2011 were used, since for the other no ft-PSICAM data were available. The optical  
405 proxy data were converted to [TSM] and [chl-*a*] using the appropriate linear equations for the  
406 respective instrument shown in table 2 (based on the non-logarithmized data). As the offset  
407 was negligible for the absorption coefficient- based relationships and it is more realistic to  
408 assume zero absorption at the absence of TSM or chl-*a*, the fits were forced through zero  
409 previously to the calculation (resulting in no alteration in the slope for TSM and a minimal  
410 change in the slope to 0.018 for chl-*a*).

411

#### 412 **3.2.1 Total suspended matter determined on transects**

413 The resulting time series for calculated [TSM] of different origin are shown in figure 5.  
414 Although the occasionally occurring negative [TSM] do not comply with reality but were a  
415 result from turbidity values below the offset, they were nevertheless included in the graph to  
416 show the relative progression of the data. In general, this progression was quite similar for  
417 the values calculated from both optical proxies, but there were also noticeable differences  
418 (e.g. 10th April, 12th April or large parts of the June cruise). Although most of the differences



419 were visible at low [TSM] where the variation of the instruments was relatively high (compare  
420 3.1.1), they appeared not be caused by simple scattering, but more likely due to systematic  
421 errors. One potential error could be related to the calculation procedure of the absorption  
422 coefficients during the transect measurements: In contrast to the stations, where reference  
423 values were measured, they were interpolated in a linear way during the transect  
424 measurements between two stations (compare 2.4.2). The assumption of a linear  
425 development of the reference values was necessary for calculation of absorption coefficients  
426 from the continuous measurements. In reality, however, it is likely that short-time fluctuations  
427 in reference light intensity occurred due to instabilities in the light source or contaminations of  
428 the cavity. Although the influence of these fluctuations is difficult to quantify, it could be an  
429 explanation for the peak-like deviations of the continuous absorption-based [TSM] (Fig. 5 A,  
430 10th-12th April). To avoid this in the future, reference measurements have to be performed  
431 more frequently.

432 In part, this might be also an explanation for some of the deviations visible in figure 5 B (June  
433 cruise). But here, some differences often regarded only the absolute values while the general  
434 progression of the data was similar between both [TSM] determination approaches. They  
435 were caused by the fact that a relationship obtained over several cruises instead of a cruise-  
436 specific one was used for calculating [TSM]. Comparing figure 2 C, it can be estimated that a  
437 fit applied only to the values of this cruise (black squares) would have a smaller slope  
438 compared to that calculated in the figure. The manually taken turbidity data from this cruise  
439 did not differ in a similar way from those of the other cruises (Fig. 2 A). Therefore, the  
440 systematic underestimation of turbidity by the SCUFA-II during this cruise indicated a  
441 problem with its calibration, or the sensor has been contaminated during the cruise.

442 Another possibility is of course that the deviations in the calculated [TSM] resulted not  
443 necessarily from problems with the sensors, but could be at least in part due to the  
444 aforementioned different optical properties of the suspended material. But at such low [TSM],  
445 where most of the deviations were observed, it is likely that potential natural variation was  
446 masked by variation due to the instrumental errors. Thus, for further comparison of the

447 turbidity- and absorption-based approach, additional data from areas with high [TSM]  
448 (estuaries, tidelands) would be useful. When such data were available, such measurements  
449 of  $a_{p,700nm}$  and turbidity in parallel would be interesting to evaluate if variations in the ratio of  
450 both proxies can be attributed to differences in the nature of the suspended material.

451

### 452 **3.2.2 Chlorophyll-a determined on transects**

453 The [chl-a] time series resulting from the conversion of the optical proxy data for the two  
454 cruises investigated are shown in figure 6 A. As for TSM, negative values for the  
455 fluorescence derived [chl-a] result from fluorescence values smaller than the offset and were  
456 retained to document the relative progression of the data.

457 The progressions of both [chl-a] datasets were generally in good agreement, but frequently  
458 the absolute values were considerably different. This can be explained due to the higher  
459 variability of fluorescence compared to chl-a absorption. To a certain extent, light  
460 acclimatization (probably by non-photochemical quenching) of the phytoplankton played a  
461 role and resulted in a reduction of the fluorescence signal at high light intensities (Abbott, et  
462 al., 1982). This led to an underestimation of the fluorescence-based [chl-a] compared to the  
463 absorption-based [chl-a]. These light-induced differences can be seen in figure 6 B, where  
464 the relation of [chl-a] derived from both optical proxies is shown together with the  
465 photosynthetically active radiation (PAR) measured outside the water for the respective  
466 period. During night, the relation was close to 1 for most of the time, indicating a comparable  
467 performance of both methods in determining [chl-a]. But when PAR increased during the day,  
468 the relation increased to values above 8, caused by a drop in the fluorescence signal. This  
469 observed fluorescence quenching at the sea surface around noon is a typical feature that  
470 alters the fluorescence / [chl-a] ratio (Kiefer, 1973a; Loftus and Seliger, 1975; Babin, et al.,  
471 1996).

472 When light availability is reduced as in the presence of high turbidity or high CDOM values  
473 (Fig. 6 C) the relationship did not or only weakly increase, although PAR is high. Thus, it can  
474 be concluded that the observed peaks were indeed light induced and not because of an

475 inherent diel rhythm of the cells. This demonstrates that even if chl-*a* fluorescence is a more  
476 variable proxy than  $a_{\Phi 676 \text{ nm}}$  with respect to [chl-*a*] determination, the simultaneous  
477 measurement of both parameters can be used to obtain additional information about the  
478 status of the phytoplankton.

479

#### 480 **4. Summary and conclusion**

481 It was shown that [TSM] and [chl-*a*], two important parameters in biological oceanography,  
482 can be reasonably well measured using optical proxy values (turbidity, fluorescence and  
483 absorption coefficients). However, especially for spatially and temporally large  
484 heterogeneities (as we have in the German Bight when conducting long-term  
485 measurements), the absorption-based approach was indicated as being more reliable. For  
486 both [TSM] and [chl-*a*] the relationship to the corresponding absorption coefficient ( $a_{p700 \text{ nm}}$   
487 and  $a_{\Phi 676 \text{ nm}}$ ) was more stable over the different seasons than it was to the respective  
488 traditional optical proxy, with the difference being more pronounced for chl-*a*. This was also  
489 the case for coefficients obtained continuously with the ft-PSICAM, although the variability  
490 was higher compared to the data obtained manually with the conventional PSICAM. The  
491 weaker performance of the turbidity and fluorescence-based approach was attributed to a  
492 higher susceptibility to variations in the investigated material. An influence of ambient PAR  
493 on [chl-*a*] determined via fluorescence was demonstrated.

494 Although allowing a more robust determination of [TSM] and [chl-*a*], continuous absorption  
495 measurements based on the PSICAM principle are by now less convenient than *in situ* chl-*a*  
496 fluorescence or turbidity measurements. Thus, for some scientific questions which do not  
497 need the higher accuracy, this approach might be too sophisticated. For other questions, e.g.  
498 the validation of remote sensing data or productivity calculations, it might be desirable and  
499 worth the effort. Furthermore, the ft-PSICAM has, since it is only a first attempt to use the  
500 PSICAM principle for continuous measurements, potential for further improvement in the  
501 future, namely achieving accuracy comparable to the high precision of conventional PSICAM  
502 measurements and a more convenient use by automation.

503 Moreover, devices like the PSICAM and the ft-PSICAM measure not only the absorption  
504 coefficients important for [chl-*a*] and [TSM] determination, but those of the whole visible  
505 spectrum. Such data could be used for the discrimination and identification of phytoplankton  
506 groups, what would be an additional set of valuable information for a variety of ecological  
507 questions. Therefore, it is important to put further effort in the improvement and development  
508 of such systems for reliable, high frequency measurement of water constituent absorption  
509 coefficients in the whole visible light spectrum.

510

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517 Biology of the Marine Environment) of the University of Oldenburg for providing PAR-data for  
518 the time of the relevant cruises. Thank goes also to the reviewers which helped to improve  
519 the quality of the manuscript.

520

521 **Table / figure captions**

522

523 Tab. 1: Time of the different cruises and number of stations sampled.

524

<b>cruise</b>	<b>date</b>	<b>no. of stations</b>
September 2010	14 <sup>th</sup> – 20 <sup>th</sup>	19
April 2011	5 <sup>th</sup> – 12 <sup>th</sup>	33
June 2011	16 <sup>th</sup> – 22 <sup>th</sup>	38
September 2011	15 <sup>th</sup> – 20 <sup>th</sup>	24

525

526

527

528 Tab. 2: Equations of linear fits shown in figure 2 and 5.  $P > 0.05$  in all cases according to t-

529 tests.

<b>device</b>	<b>equation</b>	<b>standard error of slope [%]</b>	<b>fig.</b>	<b>equation (log data)</b>	<b>standard error of slope [%]</b>	<b>fig.</b>	<b>n</b>
Hach	$y = 0.011 + 0.869x$	2.3	2A	$y = -0.097 + 1.042x$	2.7	2B	97
SCUFA-II	$y = 0.684 + 0.366x$	2.7	2C	$y = -0.033 + 0.674x$	3.5	2D	100
PSICAM	$y = 0.001 + 0.016x$	1.3	2E	$y = -1.784 + 0.987x$	2.2	2F	98
ft-PSICAM	$y = -0.001 + 0.015x$	4.2	2G	$y = -1.832 + 0.893x$	9.9	2H	42

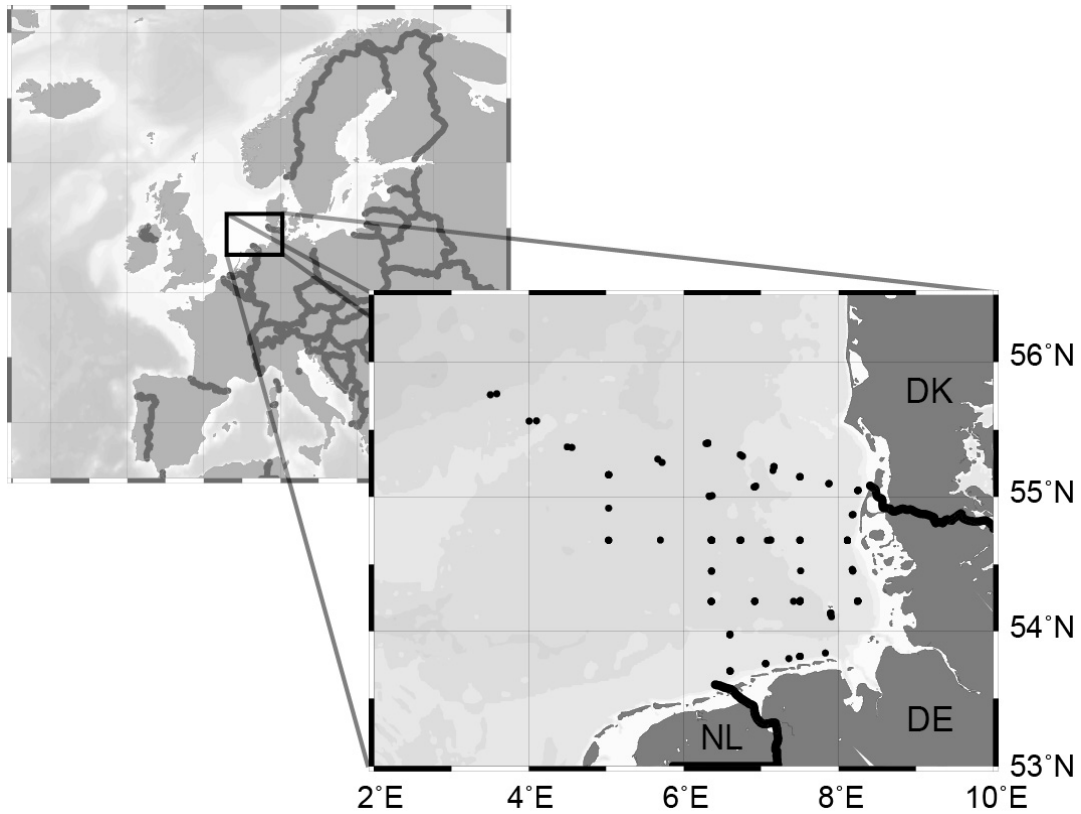
<b>device</b>	<b>equation</b>	<b>standard error of slope [%]</b>	<b>fig.</b>	<b>equation (log data)</b>	<b>standard error of slope [%]</b>	<b>fig.</b>	<b>n</b>
SCUFA-II	$y = 0.547 + 0.422x$	6.2	5A	$y = -0.023 + 0.566x$	7.8	5B	110
SCUFA-II (night)	$y = 0.649 + 0.498x$	10.6	5C	$y = 0.087 + 0.502x$	13.3	5D	21
PSICAM	$y = 0.004 + 0.017x$	3.5	5E	$y = -1.687 + 0.901x$	3.1	5F	110
ft-PSICAM	$y = 0.011 + 0.016x$	7.1	5G	$y = -1.634 + 0.844x$	5.6	5H	50

530

531

532

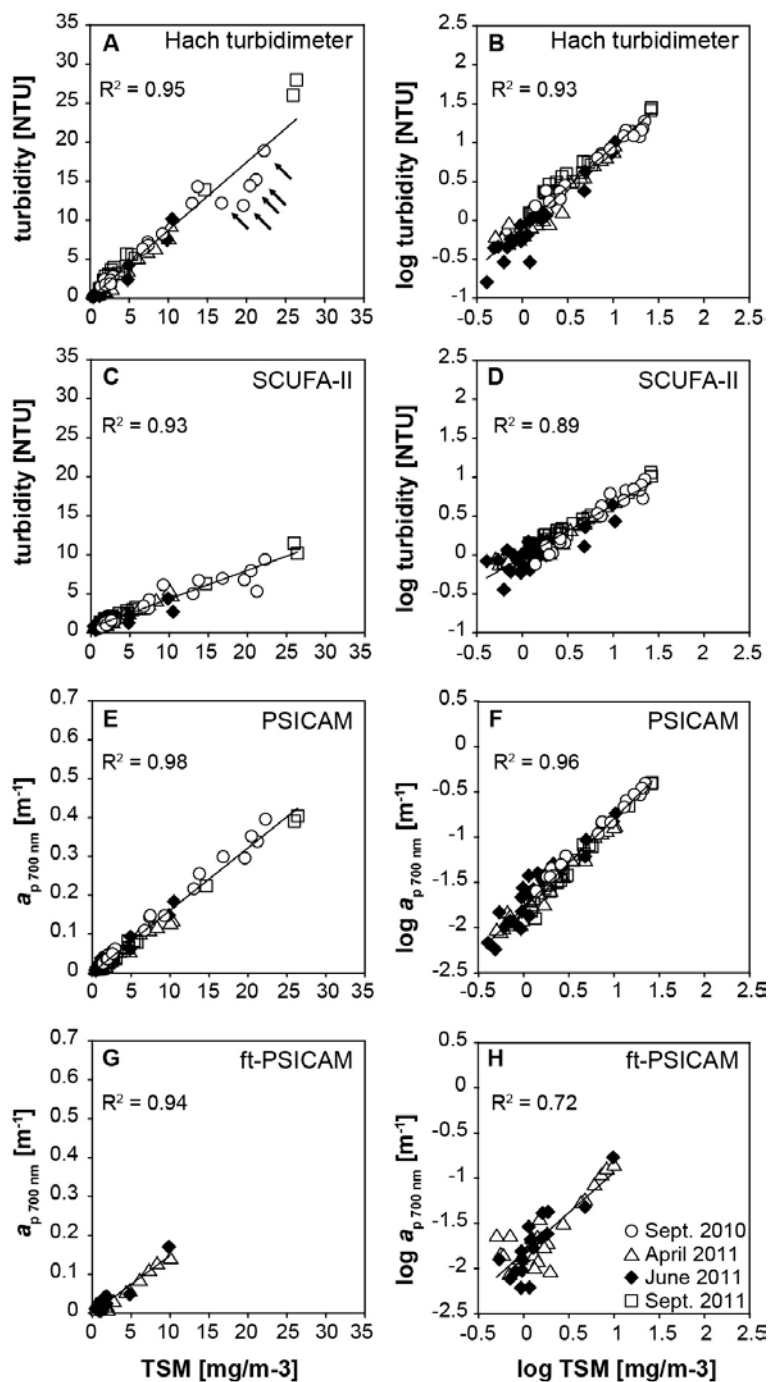
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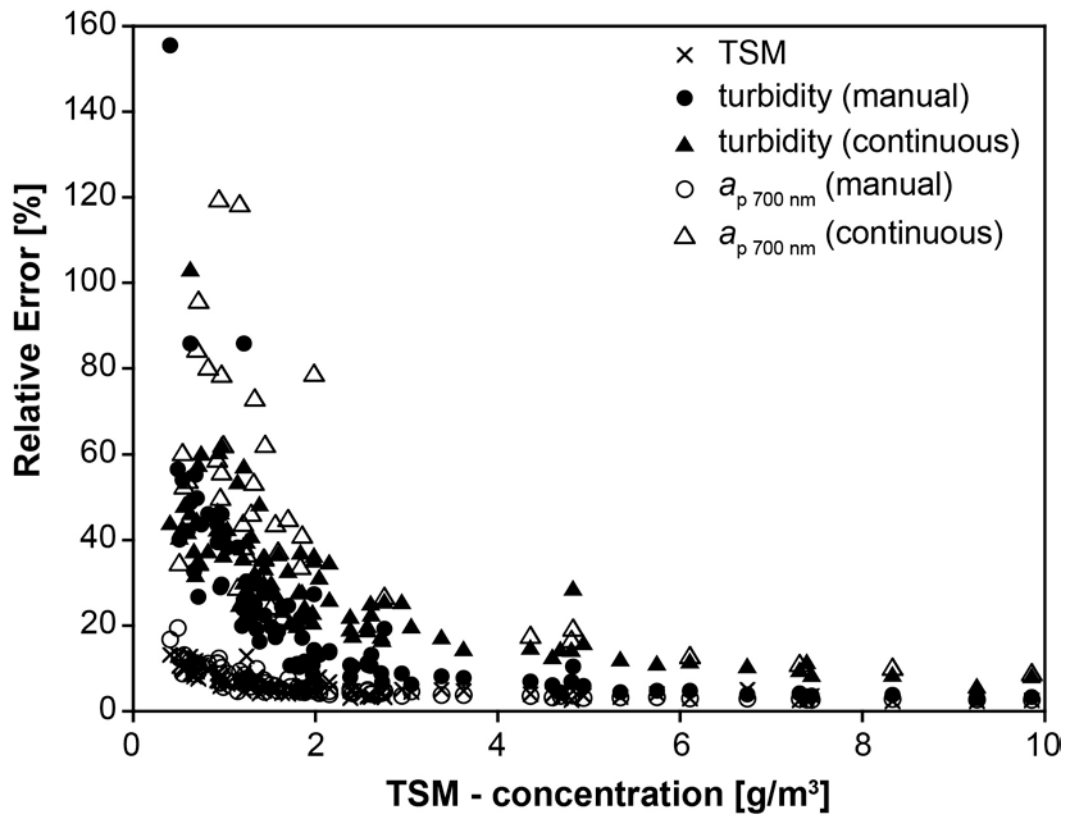
534

535 Fig. 1: Stations of the different cruises in the German Bight. Transects were carried out  
536 between the stations in different order, depending on weather conditions on the respective  
537 cruise. Different shades of grey represent the depth profile of the area (light areas = shallow,  
538 dark areas = deep). DE = Germany, DK = Denmark, NL = Netherlands. The map is based on  
539 Ocean Data View Software (Schlitzer 2012).

540



541  
 542 Fig. 2: Correlations between gravimetrically determined [TSM] and optical proxy values. (A, B)  
 543 Turbidity measured with the Hach turbidimeter, (C, D) turbidity measured with the SCUFA-II  
 544 (E, F)  $a_{p\ 700\ \text{nm}}$  obtained by the conventional PSICAM, (G, H)  $a_{p\ 700\ \text{nm}}$  obtained by the ft-  
 545 PSICAM. Note that the panels on the right side display the data as logarithmic values.  
 546 Arrows in (A) mark stations with low turbidity in relation to the measured [TSM].  
 547

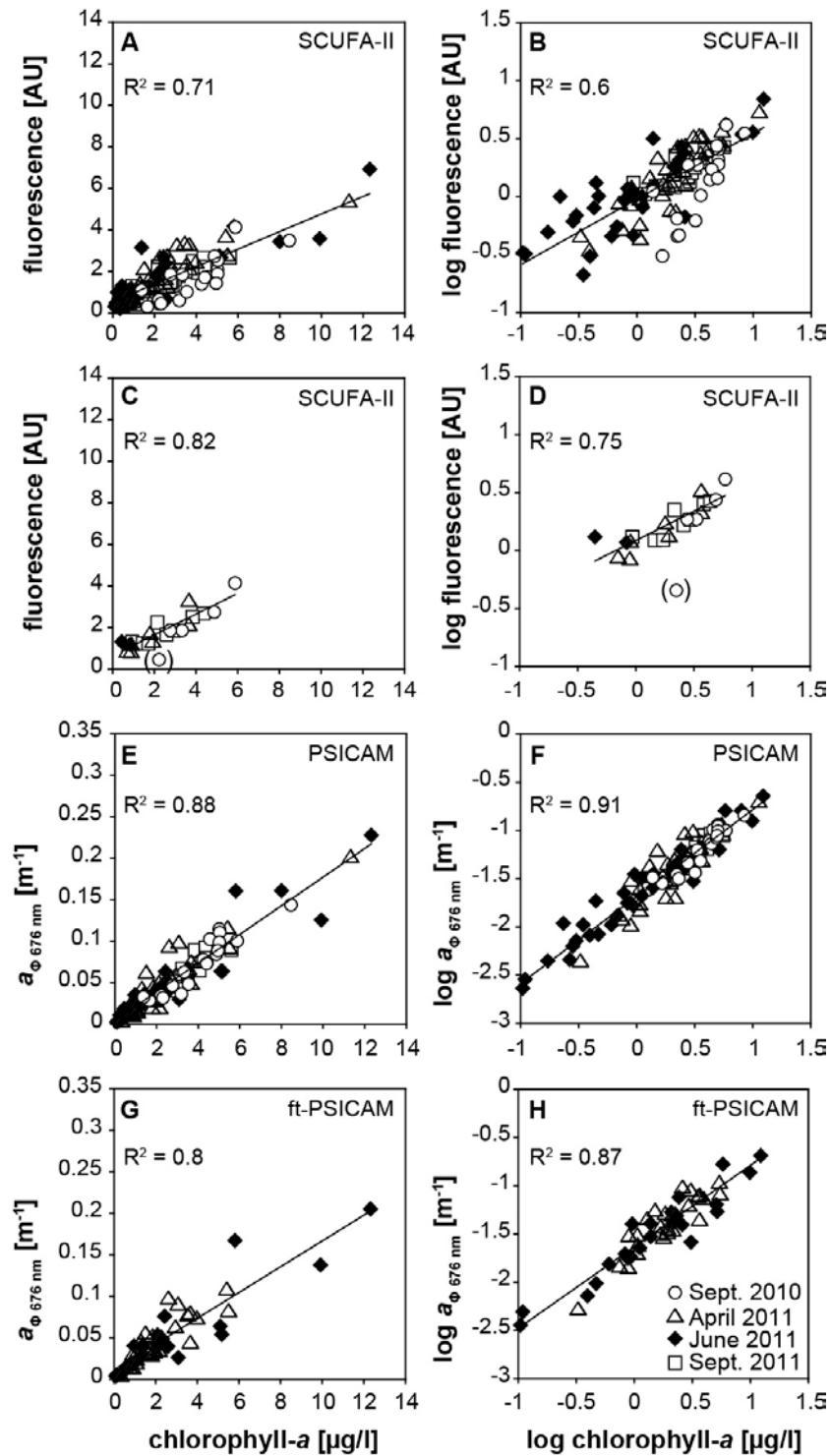


548

549 Fig. 3: Relative error of the different methods for [TSM] determination (gravimetry, turbidity  
 550 measurements and absorption measurements) as a function of [TSM].

551





552

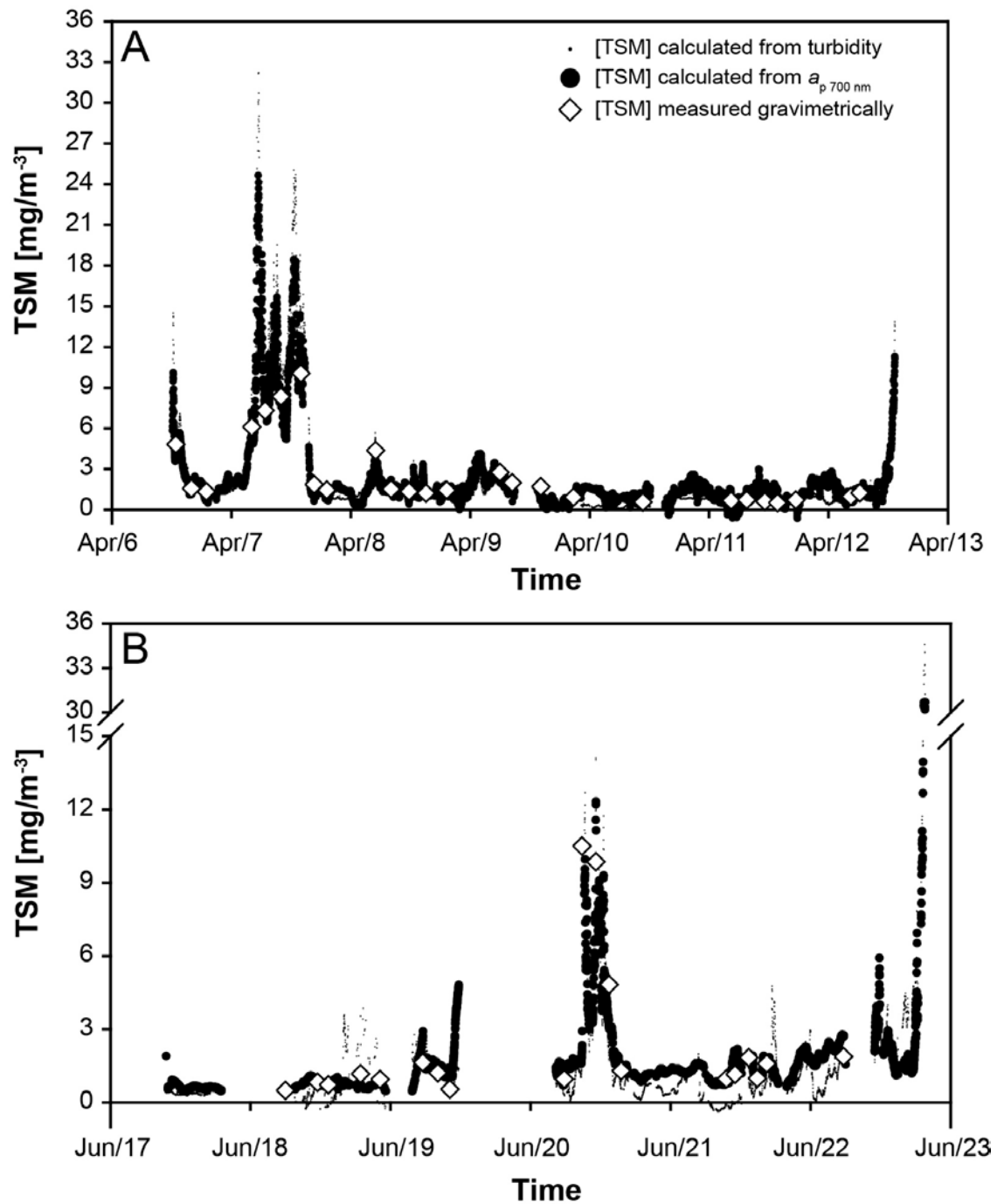
553 Fig. 4: Correlations between [chl-a] measured by HPLC and optical proxy values. (A, B)

554 Fluorescence measurements, (C, D) fluorescence measurements made after sunset, (E, F)

555  $a_{\phi 676 \text{ nm}}$  obtained by the conventional PSICAM, (G, H)  $a_{\phi 676 \text{ nm}}$  obtained by the ft-PSICAM.

556 Note that the panels on the right side display the data as logarithmic values.

557



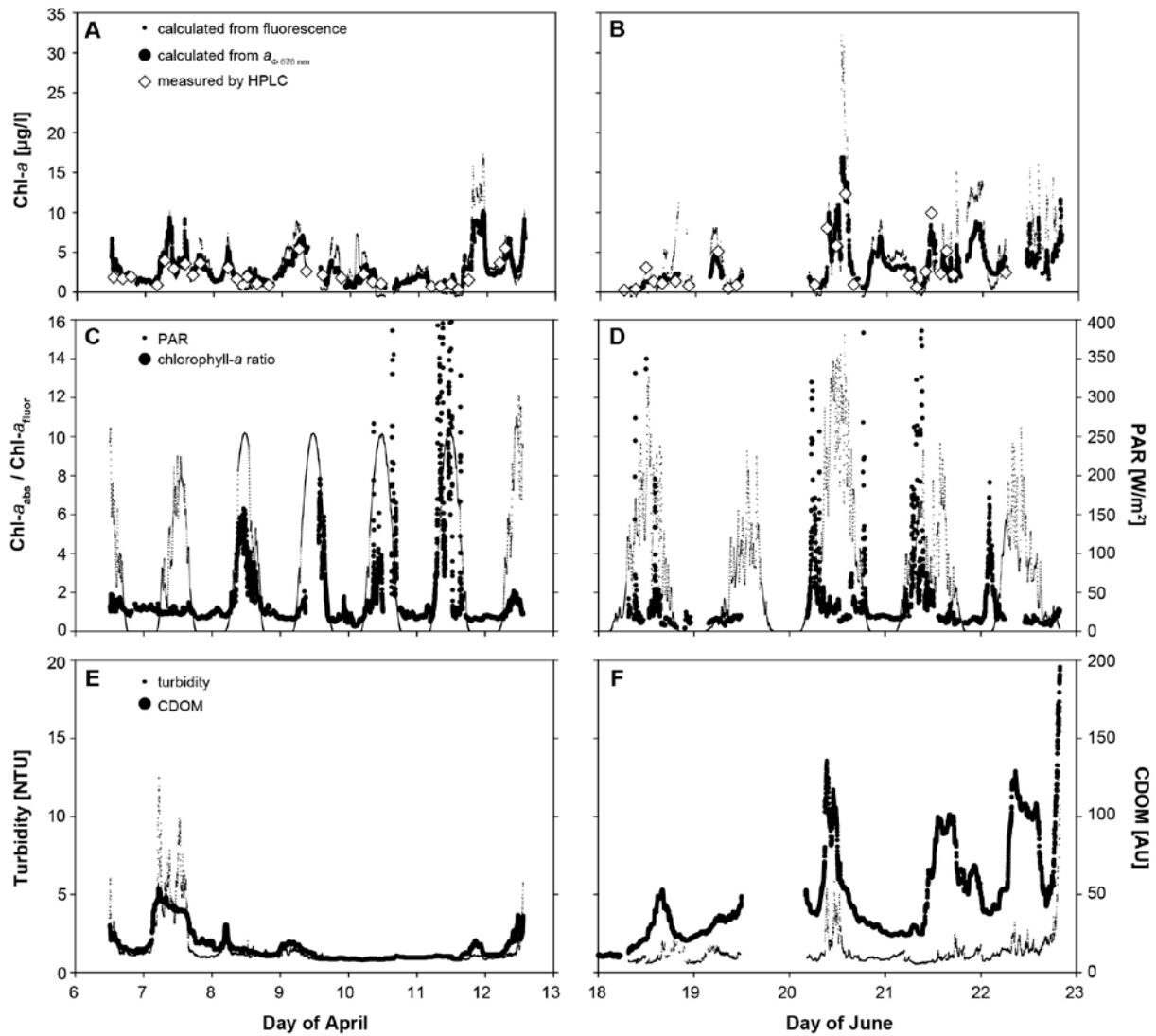
558

559 Fig. 5: Time series of [TSM] calculated from continuous turbidity and  $a_{p700\text{ nm}}$  measurements

560 taken in April 2011 (A), and June 2011 (B). Gravimetrically determined values are shown for

561 comparison.

562



563

564 Fig. 6: Time series of the cruises in April 2011 (left) and June 2011 (right). (A, B) [chl-a]  
 565 derived from fluorescence,  $a_{\Phi, 676 \text{ nm}}$  and HPLC; (C, D) Relation of [chl-a] calculated from both  
 566 proxies in dependence of PAR; (E, F) turbidity and CDOM for the respective time period. The  
 567 legends on the left panels apply also for the respective right panels.

568

569

570

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